

# Synthesis of the C22–C26 Tetrahydropyran Segment of Phorboxazole by a Stereoselective Prins Cyclization

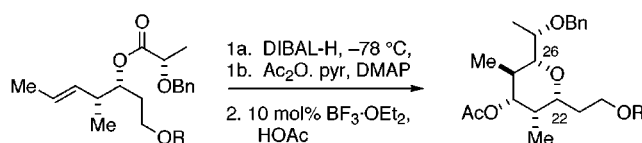
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## ABSTRACT



Tetrahydropyran rings are found in many complex natural products, and the segment-coupling Prins cyclization is an effective strategy for their synthesis. We report a four-step, stereoselective synthesis of the C20–C27 tetrahydropyran segment of phorboxazole. The key step is a Prins cyclization induced by *catalytic*  $\text{BF}_3 \cdot \text{OEt}_2$ .

Phorboxazoles A and B have attracted the attention of synthetic chemists because of their very high potency in the NCI anticancer screens (mean  $\text{GI}_{50}$   $1.6 \times 10^{-9}$  M), the difficulty of securing material from natural sources, and their interesting architecture.<sup>1</sup> Forsyth's group led the way with the first total synthesis of phorboxazole A in 1998.<sup>2</sup> A number of other groups have reported work directed toward the total syntheses of the phorboxazoles.<sup>3</sup> Our recent interest in these compounds has focused on the assembly of the four

tetrahydropyran rings present in the structures. We describe an enantioselective and stereoselective synthesis of the C22–C26 tetrahydropyran ring of phorboxazole A.

The Prins reaction between a homoallylic alcohol and an aldehyde in acidic solution is a well-established synthesis of tetrahydropyrans.<sup>4</sup> Protic and Lewis acid promoted cy-

(1) (a) Searle, P. A.; Molinski, T. F. *J. Am. Chem. Soc.* **1995**, *117*, 8126–8131. (b) Searle, P. A.; Molinski, T. F.; Brzezinski, L. J.; Leahy, J. W. *J. Am. Chem. Soc.* **1996**, *118*, 9422–9423. (c) Molinski, T. F. *Tetrahedron Lett.* **1996**, *37*, 7879–7880.

(2) (a) Forsyth, C. J.; Ahmed, F.; Cink, R. D.; Lee, C. S. *J. Am. Chem. Soc.* **1998**, *120*, 5597–5598. (b) Lee, C. S.; Forsyth, C. J. *Tetrahedron Lett.* **1996**, *37*, 6449–6452. (c) Cink, R. D.; Forsyth, C. J. *J. Org. Chem.* **1997**, *62*, 5672–5673. (d) Ahmed, F.; Forsyth, C. J. *Tetrahedron Lett.* **1998**, *39*, 183–186.

(3) (a) Ye, T.; Pattenden, G. *Tetrahedron Lett.* **1998**, *39*, 319–322. (b) Pattenden, G.; Plowright, A. T.; Tornos, J. A.; Ye, T. *Tetrahedron Lett.* **1998**, *39*, 6099–6102. (c) Paterson, I.; Arnott, E. A. *Tetrahedron Lett.* **1998**, *39*, 7185–7188. (d) Williams, D. R.; Clark, M. P.; Berliner, M. A. *Tetrahedron Lett.* **1999**, *40*, 2287–2290. (e) Williams, D. R.; Clark, M. P. *Tetrahedron Lett.* **1999**, *40*, 2291–2294. (f) Smith, A. B., III.; Verhoest, P. R.; Minbiole, K. P.; Lim, J. J. *Org. Lett.* **1999**, *1*, 909–912. (g) Smith, A. B., III.; Minbiole, K. P.; Verhoest, P. R.; Beauchamp, T. J. *Org. Lett.* **1999**, *1*, 913–916. (h) Wolbers, P.; Misske, A. M.; Hoffmann, H. M. R. *Tetrahedron Lett.* **1999**, *40*, 4527–4530. (i) Wolbers, P.; Hoffmann, H. M. R. *Synthesis* **1999**, 797–802. (j) Wolbers, P.; Hoffmann, H. M. R.

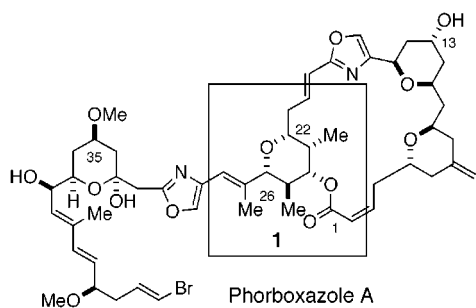
*Tetrahedron* **1999**, *55*, 1905–1914. (k) Misske, A. M.; Hoffmann, H. M. R. *Tetrahedron* **1999**, *55*, 4315–4324. (l) Evans, D. A.; Cee, V. J.; Smith, T. E.; Santiago, K. J. *Org. Lett.* **1999**, *1*, 87–90. (m) Schaus, J. V.; Panek, J. S. *Org. Lett.* **2000**, *2*, 469–471.

(4) (a) Adams, D. R.; Bhatnagar, S. P. *Synthesis* **1977**, 661–672. (b) Snider, B. B. *The Prins Reaction and Carbonyl Ene Reactions*; Trost, B. M., Fleming, I., Heathcock, C. H., Eds.; Pergamon Press: New York, 1991; Vol. 2, pp 527–561.

(5) (a) Stapp, P. R. *J. Org. Chem.* **1969**, *34*, 479–485. (b) Kay, I. T.; Williams, E. G. *Tetrahedron Lett.* **1983**, *24*, 5915–5918. (c) Kay, I. T.; Bartholomew, D. *Tetrahedron Lett.* **1984**, *25*, 2035–2038. (d) Bunnelle, W. H.; Seamon, D. W.; Mohler, D. L.; Ball, T. F.; Thompson, D. W. *Tetrahedron Lett.* **1984**, *25*, 2653–2654. (e) Winstead, R. C.; Simpson, T. H.; Lock, G. A.; Schiavelli, M. D.; Thompson, D. W. *J. Org. Chem.* **1986**, *51*, 275–277. (f) Coppi, L.; Ricci, A.; Taddie, M. *J. Org. Chem.* **1988**, *53*, 911–913. (g) Lolkema, L. D. M.; Hiemstra, H.; Mooiweer, H. H.; Speckamp, W. N. *Tetrahedron Lett.* **1988**, *29*, 6365–6368. (h) Nikolic, N. A.; Gonda, E.; Longford, C. P. D.; Lane, N. T.; Thompson, D. W. *J. Org. Chem.* **1989**, *54*, 2748–2751. (i) Mohr, P. *Tetrahedron Lett.* **1993**, *34*, 6251–6254. (j) Hu, Y.; Skalitzky, D. J.; Rychnovsky, S. D. *Tetrahedron Lett.* **1996**, *37*, 8679–8682. (k) Rychnovsky, S. D.; Yang, G.; Hu, Y. Q.; Khire, U. R. *J. Org. Chem.* **1997**, *62*, 3022–3023. (l) Yang, J.; Viswanathan, G. S.; Li, C. J. *Tetrahedron Lett.* **1999**, *40*, 1627–1630. (m) Cloninger, M. J.; Overman, L. E. *J. Am. Chem. Soc.* **1999**, *121*, 1092–1093.

clizations work well in simple cases, and these reactions are often stereoselective.<sup>5</sup> Prins cyclizations have occasionally been used in the synthesis of more complex targets.<sup>6</sup> Our recently reported reductive acetylation of esters to  $\alpha$ -acetoxy ethers<sup>7</sup> provides a facile entry into the oxocarbenium ion intermediates that are ideal precursors for segment-coupling Prins cyclizations.<sup>8</sup>

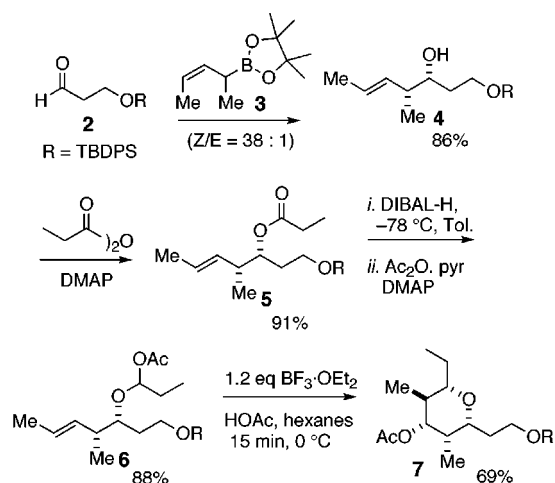
Segment **1**, which includes the C20–C26 tetrahydropyran of the phorboxazoles, is shown in Figure 1. There are a



**Figure 1.** The C22–C26 tetrahydropyran segment of phorboxazole A.

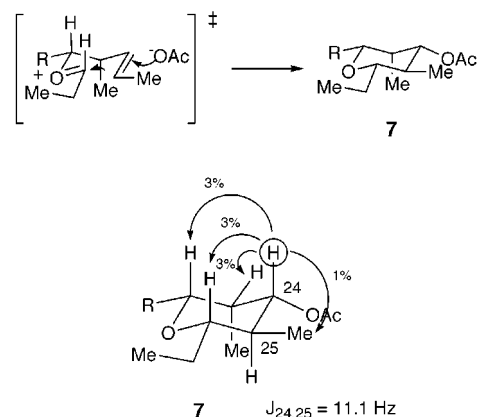
number of possible Prins cyclization precursors for **1**, but none of them are ideal for our strategy because they either involve (a) reductive acetylation of an  $\alpha,\beta$ -unsaturated ester or (b) Prins cyclization of an allylic alcohol that might be expected to fragment. To avoid these potential problems, we settled on a strategy in which the trisubstituted alkene would be incorporated after the Prins cyclization. The key elements of this approach were explored in the model study shown in Scheme 1. Aldehyde **2** was coupled with Hoffmann's (*Z*-

**Scheme 1.** Prins Cyclization in a Model System



pentenyl boronate **3**<sup>9</sup> to produce homoallylic alcohol **4**.<sup>10</sup> Alcohol **4** incorporates the requisite syn stereochemistry and the (*E*)-alkene geometry that should lead to the correct

configuration in the tetrahydropyran. Esterification and reductive acetylation proceeded uneventfully to give  $\alpha$ -acetoxy ether **6** in 80% overall yield from alcohol **4**. Activation with  $\text{BF}_3 \cdot \text{OEt}_2$  and acetic acid in hexanes at 0 °C gave tetrahydropyran **7** as a single isomer in 69% yield. The configuration of **7** was confirmed by NOE measurements and coupling constant analysis (Figure 2). A chair conforma-

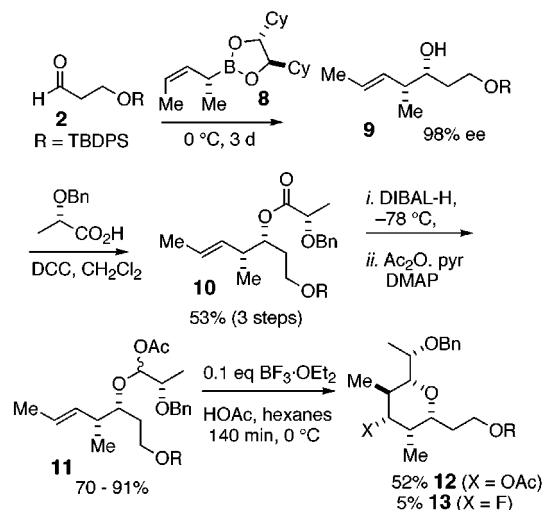


**Figure 2.** Geometry of the Prins cyclization transition state leading to tetrahydropyran **7**; NOE data and coupling constants supports the assignment.

tion of the oxocarbenium ion intermediate and a trans addition across the alkene are consistent with the observed configuration of the product, Figure 2. The successful synthesis of tetrahydropyran **7**, which incorporates the five contiguous stereogenic centers of **1**, led us to investigate the optically active substrate.

A viable precursor to the C20–C26 tetrahydropyran segment of phorboxazole would need to be optically pure and need to incorporate a functional handle for the introduc-

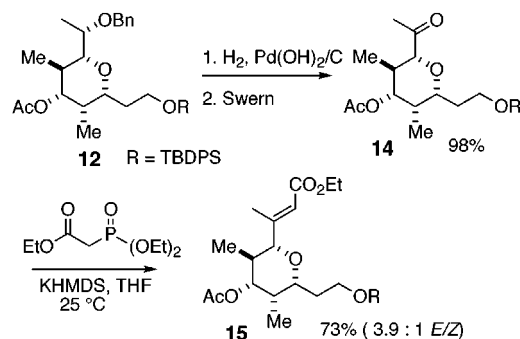
**Scheme 2.** Synthesis of the Optically Pure Phorboxazole Tetrahydropyran **12**



tion of the C27–C28 trisubstituted alkene. Tetrahydropyran **12** fulfills these requirements, and our approach to **12** is shown in Scheme 2. Aldehyde **2** was coupled with Hoffmann's optically pure (*Z*)-pentenyl boronate **8**<sup>11</sup> to give alcohol **9** with 98% ee.<sup>12</sup> A DCC coupling with *O*-benzyl *S*-lactic acid led to ester **10** as a single diastereomer. The yield of **10** was 53% overall from 3-*O*-(*tert*-butyldiphenylsilyl)propane-1,3-diol, the precursor to aldehyde **2**. Reductive acetylation of **11** worked well on a 100 mg scale and gave  $\alpha$ -acetoxy ether **11** in 91% yield. On a 10 g scale, the yield was 70%, and the product was accompanied by 18% of the overreduced product. Normally reductive acetylations scale well,<sup>7</sup> and in this case the problem was probably a too rapid addition of DIBAL-H that led to overreduction. Initial cyclization experiments on a diastereomer of **11** produced significant amount of fluoride trapping along with the desired product. We found that the use of catalytic BF<sub>3</sub>·OEt<sub>2</sub> with acetic acid largely alleviated this problem. Cyclization of  $\alpha$ -acetoxy ether **11** with 10 mol % of Lewis acid and 5 equiv of acetic acid gave the desired tetrahydropyran **12** in 52% yield, accompanied by 5% of the corresponding fluoride. This is the first example of which we are aware where a Prins cyclization was *catalyzed* by a Lewis acid. Compound **12** incorporates the five contiguous centers of the C20–C26 ring of phorbosazole and was prepared in four steps and 25% overall yield from aldehyde **2**.

Introduction of the trisubstituted alkene was evaluated as illustrated in Scheme 3. The benzyl ether of **12** was removed

**Scheme 3.** Synthesis of the Trisubstituted Alkene



by hydrogenation, and the resulting alcohol was oxidized using Swern's conditions.<sup>13</sup> Ketone **14** was isolated as a

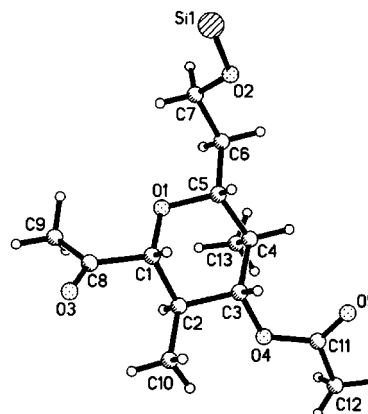
(6) For example, see: Overman, L. E.; Thompson, A. S. *J. Am. Chem. Soc.* **1988**, *110*, 2248–2256.

(7) (a) Dahanukar, V. H.; Rychnovsky, S. D. *J. Org. Chem.* **1996**, *61*, 8317–8320. (b) Kopecky, D. J.; Rychnovsky, S. D. *J. Org. Chem.* **2000**, *65*, 191–198.

(8) Rychnovsky, S. D.; Hu, Y. Q.; Ellsworth, B. *Tetrahedron Lett.* **1998**, *39*, 7271–7274.

(9) Andersen, M.; Hildebrandt, B.; Köster, G.; Hoffmann, R. W. *Chem. Ber.* **1989**, *122*, 1777–82.

crystalline solid, mp = 88 °C. X-ray analysis confirmed the relative and absolute configuration of **14**; the structure is shown in Figure 3.<sup>14</sup> Horner–Emmons reaction with the



**Figure 3.** Crystal structure of ketone **14** is shown; the silyl substituents have been omitted for clarity.

triethyl phosphonoacetate anion gave alkene **15** in 73% yield. The desired (*E*)-alkene predominated in a 3.9:1 ratio. Ester **14** has all of the key features of the C20–C26 tetrahydropyran segment **1** from phorbosazole A except for the oxazole ring.

We have described a practical and enantioselective synthesis of the C20–C26 tetrahydropyran of phorbosazole A. This work demonstrates that the segment-coupling Prins cyclization is suitable for the synthesis of highly substituted tetrahydropyran rings.

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**Supporting Information Available:** Structural information for ketone **14** as a crystallographic information file. Experimental procedures for the preparation of **15**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(10) All new compounds were characterized by <sup>1</sup>H and <sup>13</sup>C NMR, IR, and MS analysis. All compounds except **9**, which was contaminated with starting material, gave satisfactory elemental analyses.

(11) (a) Hoffmann, R. W.; Ditrich, K.; Köster, G.; Stürmer, R. *Chem. Ber.* **1989**, *122*, 1783–9. (b) Hiscox, W. C.; Matteson, D. S. *J. Org. Chem.* **1996**, *61*, 8315–8316.

(12) Alcohol **9** was analyzed for optical purity by using a Chiralcel OD-H column with 98:2 hexane:*i*-PrOH as the eluent, using the racemate as a standard. The 3*R*,4*R* enantiomer (**9**) eluted first.

(13) Mancuso, A. J.; Huang, S.-L.; Swern, D. *J. Org. Chem.* **1978**, *43*, 2480–2482.

(14) The X-ray data are available as Supporting Information. We thank Dr. Joe Ziller for determining the structure of **14**.